

**Aerplysinin-2, a Dibromolactone from Marine Sponges *Aplysina* (*Verongia*)
aerophoba and *Ianthella* sp.**

By L. MINALE* and G. SODANO

(Laboratorio per la Chimica e Fisica di Molecole di Interesse Biologico del C.N.R., Via Toiano, 2 Arco Felice, Naples, Italy)

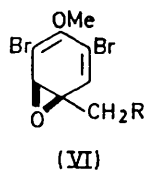
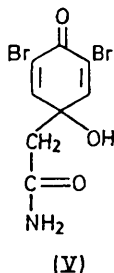
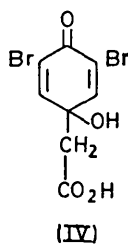
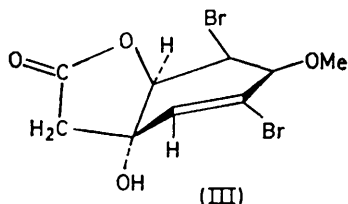
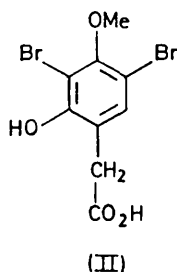
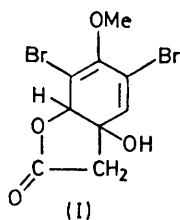
and W. R. CHAN and A. M. CHEN

(Chemistry Department, University of The West Indies, Kingston 7, Jamaica)

Summary Absolute structure (VIII) has been assigned to aerplysinin-2, a new bromo-compound from the sponges *Aplysina* (*Verongia*) *aerophoba* and *Ianthella* sp.

RECENTLY,¹⁻⁴ a number of related bromo-compounds have been isolated from *Verongia* spp. sponges; in addition, both enantiomeric isomers of one of these have been isolated

from *Ianthella* spp.⁵ After further examination of the solvent extracts from *Verongia (Aplysina) aerophoba* and *Ianthella* sp., we have now isolated a related minor dibromo-constituent, aeroplysinin-2 (I).



Aeroplysinin-2 (I), $C_9H_8Br_2O_4$ (elemental analysis and mass spectrum), has been obtained as the optically active form, m.p. 106–108°, $[\alpha] + 22^\circ$ (*c* 3 in MeOH) from *Verongia aerophoba* and as the racemate, m.p. 127–128°,

from *Ianthella* sp. It shows u.v. absorption at 284 nm [ϵ 4900 in MeOH; *cisoid* diene; cf. aeroplysinin-1³], i.r. bands ($CHCl_3$) at 3400br (OH) and 1785 (CO, γ -lactone) cm^{-1} and n.m.r. signals at δ ($CDCl_3$) 2.88 (2H, ABq, *J* 18 Hz, CH_2), 3.38br (1H, OH), 3.79 (3H, s, OMe), 5.20 (1H, d, *J* 0.7 Hz, CHO), and 6.36 (1H, d, *J* 0.7 Hz, C=CH) p.p.m.

On acetylation (I) afforded a monoacetate (oil), *m/e* 380, 382, 384 (M^+), 320, 322, and 324 (strong, $M^+ - AcOH$); ν_{max} (liquid film) 1795 (γ -lactone-CO) and 1745 (acetyl-CO); δ 2.10 ($MeCO_2$) p.p.m., which, on treatment with cold dilute alkali, gave the acetic acid (II), identified by comparison of its methyl ester with an authentic sample.³ This result, along with the spectral evidence, leads to the structure (I) for aeroplysinin-2. We propose that its stereochemistry is as represented in structure (III) for the following reasons.† Dreiding models show that the hydroxy- and acyloxy-groups must be *trans*. The small coupling (0.7 Hz) between the olefinic and methine protons suggests a **W**⁶ relationship for these two protons, indicating a quasi-equatorial orientation for the methine proton and accordingly, a quasi-diaxial orientation for the hydroxy- and acyloxy-groups. The c.d. curve [(MeOH), $\Delta\epsilon$ 284 \pm 2] indicated a right-handed helicity for the diene⁷ and, therefore, confirms the absolute configuration as depicted in (III).

When set aside, aeroplysinin-2 sometimes yielded the hydroxy-dienoic acid (IV) (identified by direct comparison with an authentic sample²), probably by a photochemical reaction. Such an easy transformation suggests that the natural amide (V) could be derived biogenetically from the lactone (I) or more directly from an epoxide (VI) which seems a likely precursor^{3,8} of aeroplysinin-2 and of all the *Verongia* bromo-metabolites.

We thank Professor M. Sará (University of Genoa) for identifying the sponges. Two of us (L.M. and G.S.) also thank the Zoological Station (Naples) for collection of *Verongia aerophoba* samples, Professor G. Geraci (Laboratorio di Embriologia Molecolare del C.N.R., Naples) for c.d. measurements, and Mr. A. Milone for mass measurements.

(Received, 13th March, 1972; Com. 420.)

† Fulmor *et al.*,⁵ following the same criteria, assigned the absolute configuration of the (–)-isomer of aeroplysinin-I; this configuration was later confirmed by *X*-ray analysis (D. B. Cosulich and F. M. Lovell, *Chem. Comm.*, 1971, 397.)

¹ G. M. Sharma and P. R. Burkholder, *Tetrahedron Letters*, 1967, 4147.

² G. M. Sharma, B. Vig, and P. R. Burkholder, *J. Org. Chem.*, 1970, **35**, 2823.

³ E. Fattorusso, L. Minale, and G. Sodano, *J.C.S. Perkin I*, 1972, 16.

⁴ K. Moody, R. H. Thomson, E. Fattorusso, L. Minale, and G. Sodano, *J.C.S. Perkin I*, 1972, 18.

⁵ W. Fulmor, G. C. Van Lear, G. O. Morton, and R. D. Mills, *Tetrahedron Letters*, 1970, 4551.

⁶ E. W. Garbisch, jun., *Chem. and Ind.*, 1964, 1715.

⁷ P. Crabbé, Applications de la dispersion rotatoire optique et du dichroïsme circulaire optique en chimie organique, Gauthier-Villars, Paris, 1968.

⁸ A. M. Reiner and G. D. Hegeman, *Biochem.*, 1971, **10**, 2530.